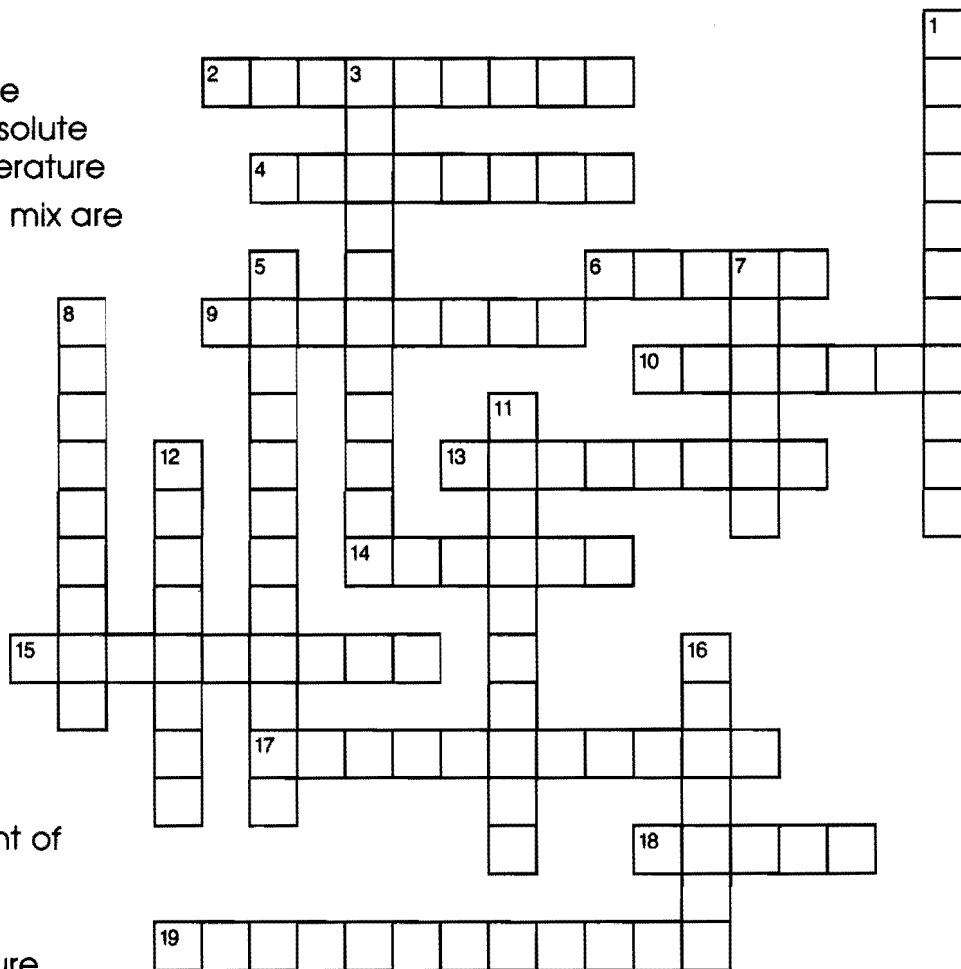


SOLUTIONS CROSSWORD

Name _____

Across

2. Solution containing the maximum amount of solute possible at that temperature
4. Two liquids which can mix are said to be _____.
6. The presence of a nonvolatile solute will _____ the boiling point of a solvent.
9. A homogeneous mixture
10. Substance present in larger amount in a mixture
13. Moles of a solute per kilogram of solvent
14. Solution containing a relatively large amount of solvent
15. The solubility of gases _____ as temperature increases.
17. State in which the rate of dissolving is equal to the rate of precipitation
18. The presence of a nonvolatile solute will _____ the freezing point of a solvent.
19. These substances dissociate or ionize in water and are then able to conduct an electric current.



Down

1. Properties that depend on the number of particles in solution
3. Solution in which more solute can be dissolved
5. Solution containing a relatively large amount of dissolved solute
7. Substance present in smaller amount in a mixture
8. The solubility of most solids _____ as temperature increases.
11. Maximum amount of solute that can dissolve in a stated amount of solute at a given temperature
12. Moles of solute per liter of solution
16. Solutions in which water is the solvent are called _____.